

# Relationship of proton motive force and the $F_0F_1$ -ATPase with bio-hydrogen production activity of *Rhodobacter sphaeroides*: effects of diphenylene iodonium, hydrogenase inhibitor, and its solvent dimethylsulphoxide

Lilit Hakobyan · Lilit Gabrielyan · Armen Trchounian

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**Abstract** *Rhodobacter sphaeroides* MDC 6521 was able to produce bio-hydrogen ( $H_2$ ) in anaerobic conditions under illumination. In this study the effects of the hydrogenase inhibitor—diphenylene iodonium ( $Ph_2I$ ) and its solvent dimethylsulphoxide (DMSO) on growth characteristics and  $H_2$  production by *R. sphaeroides* were investigated. The results point out the concentration dependent DMSO effect: in the presence of 10 mM DMSO  $H_2$  yield was ~6 fold lower than that of the control. The bacterium was unable to produce  $H_2$  in the presence of  $Ph_2I$ . In order to examine the mediatory role of proton motive force ( $\Delta p$ ) or the  $F_0F_1$ -ATPase in  $H_2$  production by *R. sphaeroides*, the effects of  $Ph_2I$  and DMSO on  $\Delta p$  and its components (membrane potential ( $\Delta\psi$ ) and transmembrane pH gradient), and ATPase activity were determined. In these conditions  $\Delta\psi$  was of  $-98$  mV and the reversed  $\Delta pH$  was  $+30$  mV, resulting in  $\Delta p$  of  $-68$  mV.  $Ph_2I$  decreased  $\Delta\psi$  in concentrations of  $20$   $\mu M$  and higher; lower concentrations of  $Ph_2I$  as DMSO had no valuable effect on  $\Delta\psi$ . The *R. sphaeroides* membrane vesicles demonstrated significant ATPase activity sensitive to *N,N'*-dicyclohexylcarbodiimide. The  $10$ – $20$   $\mu M$   $Ph_2I$  did not affect the ATPase activity, whereas  $40$   $\mu M$   $Ph_2I$  caused a marked inhibition (~2 fold) in ATPase activity. The obtained results provide novel evidence on the involvement of hydrogenase and the  $F_0F_1$ -ATPase in  $H_2$

production by *R. sphaeroides*. Moreover, these data indicate the role of hydrogenase and the  $F_0F_1$ -ATPase in  $\Delta p$  generation. In addition, DMSO might increase an interaction of nitrogenase with  $CO_2$ , decreasing nitrogenase activity and affecting  $H_2$  production.

**Keywords** Proton motive force · ATPase activity · Bio-hydrogen production · Diphenylene iodonium · Dimethylsulphoxide · *Rhodobacter sphaeroides*

## Introduction

Phototrophic microorganisms have been widely used in the energy conservation mechanism studies (Gabrielyan and Trchounian 2009a; Kapdan and Kargi 2006). The purple bacteria are perspective objects for biological hydrogen ( $H_2$ ) production, which is connected with the absorption of light energy and hence can increase the efficiency of use of the solar radiation (Gabrielyan and Trchounian 2009a; Kapdan and Kargi 2006). These bacteria can use a wide variety of substrates for growth and  $H_2$  production (Gabrielyan and Trchounian 2009a; Kapdan and Kargi 2006; Gabrielyan et al. 2010; Gabrielyan and Trchounian 2012; Hakobyan et al. 2012).

$H_2$  metabolism in phototrophic bacteria involves a coordinated action of two enzymes: nitrogenase and hydrogenase. However, the yield of  $H_2$  in hydrogenase-catalyzed reaction is much higher than that of nitrogenase-catalyzed reaction (Gabrielyan and Trchounian 2009a; Basak and Das 2007; Das et al. 2006; Hallenbeck 2009). Photoproduction of  $H_2$  by the phototrophic bacteria is suggested to be a nitrogenase-dependent reaction, whereas hydrogenases may be active for  $H_2$  uptake in many photosynthetic bacteria; but it

L. Hakobyan · L. Gabrielyan · A. Trchounian (✉)  
Department of Biophysics, Yerevan State University,  
0025 Yerevan, Armenia  
e-mail: Trchounian@ysu.am

A. Trchounian  
Department of Microbiology & Microbe and Plant Biotechnology,  
Biology Faculty, Yerevan State University,  
0025 Yerevan, Armenia

depends on different factors (Kapdan and Kargi 2006; Basak and Das 2007; Das et al. 2006; Hallenbeck 2009).

According to the chemiosmotic theory of energetic coupling, the proton motive force ( $\Delta p$ ) across the cytoplasmic membrane plays a key role in transformation of biological energy. It can be formed by carrier-mediated excretion of fermentation end products in the symport with  $H^+$  and utilized with ATP synthesis (Abee et al. 1988; Boyer 1988). Also, in certain bacteria, the reaction can be reversed and the  $F_0F_1$ -ATPase can generate  $\Delta p$ , working as an  $H^+$ -pumping ATPase (Feniouk et al. 2007). The mechanisms for energy transformation in purple non-sulfur bacterium *Rhodobacter sphaeroides* are not clear. Moreover, these bacteria produce  $H_2$  under reducing conditions upon the drop in oxidation-reduction (redox) potential ( $E_h$ ), which could determine electron transfer within a bacterial membrane and generation of  $\Delta p$  (Hakobyan et al. 2011). A relationship between pH,  $E_h$ ,  $\Delta p$  and  $H_2$  production is proposed for these bacteria, but its nature is unclear.

Various inhibitors suppress both  $H_2$  uptake and production by purple bacteria (Gabrielyan and Trchounian 2009a). Magnani with co-workers (Magnani et al. 2000) have shown inhibition of *Rhodobacter capsulatus* hydrogenase by diphenylene iodonium ( $Ph_2I$ ), a known inhibitor of mitochondrial complex I and various monooxygenases (Gatley and Sheratt 1976; Ellis et al. 1988; Doussiere and Vignais 1992). The action mode of  $Ph_2I$  on hydrogenase is not fully clarified, but a study of the effect of  $Ph_2I$  on  $H_2$  production by purple bacteria can be helpful for identifying of responsible enzymes, understanding of mechanisms and pathways of  $H_2$  production.

As known,  $Ph_2I$  is dissolved in a water-miscible solvent such as dimethylsulphoxide (DMSO), which has wide applications in cell biology and biochemistry. DMSO itself affects the membrane stability (Gordeliy et al. 1998; Yu and Quinn 1998), and bacterial growth properties (Horne and McEwan 1998; Markarian et al. 2002). DMSO at low concentration serves as an electron donor for an anaerobic respiratory chain stimulating *Escherichia coli* growth and survival through the modulation the  $H^+$ - $K^+$ -exchange and  $H_2$  production activity (Markarian et al. 2002).

In our previous studies, we have shown that *R. sphaeroides* strains isolated from mineral springs in Armenia were able to grow and produce  $H_2$  in anaerobic conditions upon illumination using various carbon and nitrogen sources (Gabrielyan et al. 2010; Gabrielyan and Trchounian 2012; Hakobyan et al. 2012). To investigate the role of  $\Delta p$  in  $H_2$  production by *R. sphaeroides*,  $\Delta p$  and its components (the membrane potential ( $\Delta\psi$ ) and the transmembrane pH gradient ( $\Delta pH$ )) were determined (Hakobyan et al. 2011).

In this study the effect of  $Ph_2I$  and its solvent DMSO on growth properties (such as specific growth rate), external medium characteristics (pH,  $E_h$ , etc.) and  $H_2$  production by

*R. sphaeroides* str. MDC 6521 was investigated. In order to examine the mediatory role of  $\Delta p$  or the  $F_0F_1$ -ATPase in  $H_2$  production by *R. sphaeroides*, the effects of  $Ph_2I$  on  $\Delta p$  and its components ( $\Delta\psi$  and  $\Delta pH$ ), and ATPase activity was studied. This study was, therefore, important for understanding of mechanisms for biological effects of DMSO and  $Ph_2I$  on bacterial growth properties and  $H_2$  production, and for revealing a role of hydrogenase in these processes.

## Materials and methods

### Bacterial strain and growth conditions

The purple non-sulfur bacterium *R. sphaeroides* MDC 6521 (Microbial Depository Center, Armenia, WDCM803) used in this study was isolated from Arzni mineral springs in Armenia (Gabrielyan and Trchounian 2009b).

*R. sphaeroides* MDC 6521 was grown in anaerobic conditions in Ormerod medium at  $30 \pm 2^\circ C$  under illumination of approximately 1500 lux (Gabrielyan et al. 2010; Gabrielyan and Trchounian 2009b). Succinate ( $3.54 \text{ g L}^{-1}$ ) and yeast extract ( $2 \text{ g L}^{-1}$ ) were used as sole carbon and nitrogen sources (Hakobyan et al. 2012). The concentration of  $Ph_2I$  ranged from 1, 5, 10, 20 to 40  $\mu M$ ; the concentration of DMSO in the growth medium ranged from 10, 30 to 50 mM.

The medium initial pH was adjusted to 7.0 by means of 0.1 M HCl or NaOH. The pH of growth medium was measured at certain time intervals (0–96 h) by a pH-potentiometer (HANNA Instruments, Portugal) with selective pH electrode (HJ1131B) (Gabrielyan et al. 2010; Gabrielyan and Trchounian 2012; Hakobyan et al. 2012; Gabrielyan and Trchounian 2009b). The bacterial growth was monitored by a Spectro UV-Vis Auto spectrophotometer (Labomed, USA) by changes in absorbance of cell suspension at 660 nm (Gabrielyan and Trchounian 2009b). Specific growth rate ( $h^{-1}$ ) was determined as described (Gabrielyan et al. 2010; Gabrielyan and Trchounian 2012; Hakobyan et al. 2012). The absorption spectra of *R. sphaeroides* suspension were recorded at the wavelength region of 400 to 1000 nm on a programmable Spectro UV-Vis Auto spectrophotometer (Labomed, USA) (Gabrielyan and Trchounian 2012; Hakobyan et al. 2012). For obtaining comparable data the original spectra were subtracted of the scattering and normalized to the same cell concentration.

### Redox potential determination and hydrogen production assays

The redox potential ( $E_h$ ) of bacterial growth medium was measured at certain time intervals (0–96 h) by using a platinum (Pt) electrode (EPB-1, Gomel State Enterprise of

Electrometric Equipment (GSEEE), Gomel, Belarus, or PT42BNC, HANNA Instruments, Portugal) (Gabrielyan and Trchounian 2012; Hakobyan et al. 2012; Trchounian et al. 2011a).

H<sub>2</sub> production was assayed by using a pair of redox Pt and titanium-silicate (Ti-Si) (EO-02, GSEEE) electrodes as described (Gabrielyan and Trchounian 2012; Hakobyan et al. 2012; Trchounian et al. 2011a). This determination is close to the method using a Clark-type electrode employed by different authors (Eltsova et al. 2010; Piskarev et al. 2010): a strong correlation between  $E_h$  and H<sub>2</sub> production yield in liquid media is shown. The H<sub>2</sub> yield was calculated by the decrease of  $E_h$  to low negative values as described by Piskarev and co-workers (Piskarev et al. 2010) and expressed in mmol L<sup>-1</sup>.

H<sub>2</sub> evolution was confirmed chemically as described (Gabrielyan and Trchounian 2012; Trchounian et al. 2011a). This method was suggested for detecting enhanced H<sub>2</sub> production (Maeda and Wood 2008).

#### Determination of proton motive force

$\Delta p$  was calculated as a sum  $\Delta\psi$  and  $\Delta pH$  according to  $\Delta\mu_{H^+}/F = \Delta\psi - Z\Delta pH$  (Zakharyan and Trchounian 2001), where  $Z$  is  $RT/F$  equal to 60 mV at 30 °C.  $\Delta\psi$  was determined from the tetraphenylphosphonium cation (TPP<sup>+</sup>) distribution between the cytoplasm and the external medium as described (Hakobyan et al. 2011; Zakharyan and Trchounian 2001; Grinus et al. 1980). 1  $\mu$ M TPP<sup>+</sup> was added in the assay medium, containing 150 mM Tris-phosphate buffer, then the bacteria were added and changes of the concentration of this probe was determined by using a TPP<sup>+</sup>-selective electrode as described (Hakobyan et al. 2011; Zakharyan and Trchounian 2001; Grinus et al. 1980). The intracellular volume of  $3.75 \cdot 10^{-9}$   $\mu$ l cell<sup>-1</sup> was employed. The  $\Delta pH$  was calculated from the external ([pH]<sub>out</sub>) and intracellular ([pH]<sub>in</sub>) pH values. The [pH]<sub>in</sub> was determined by the quenching of fluorescence of 9-aminoacridine (9-AA) as described (Hakobyan et al. 2011; Zakharyan and Trchounian 2001; Puchkov et al. 1983), which was measured with a Cary Eclipse spectrofluorimeter (Varian, USA) with excitation at 390 nm and emission at 460 nm. The uptake of 9-AA by bacterial cells was determined from the disappearance of 9-AA from the assay media. For protonophore carbonyl cyanide *m*-chlorophenylhydrazone (CCCP) effects studies, cells were incubated with the reagent (2  $\mu$ M) for 10 min.

#### ATPase activity assay

Bacterial membrane vesicles were obtained from lysozyme treated cells as described (Gabrielyan and Trchounian 2009b; Trchounian and Vassilian 1994). ATPase activity in the membrane vesicles was measured by the liberation of

inorganic phosphate (P<sub>in</sub>) in the reaction with ATP by the method of Taussky and Shorr using a spectrophotometer Spectro UV-Vis Auto (Labomed, USA) (Gabrielyan and Trchounian 2009b; Trchounian and Vassilian 1994). ATPase activity was expressed in  $\mu$ M P<sub>in</sub> min<sup>-1</sup>  $\mu$ g<sup>-1</sup> protein. Protein was determined by the Lowry method (Gabrielyan and Trchounian 2009b). For *N,N'*-dicyclohexylcarbodiimide (DCCD) inhibition studies, cells were incubated with the reagent (0.5 mM) for 10 min.

#### Reagents and data processing

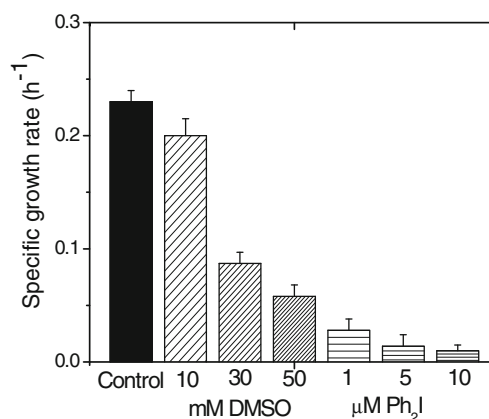
Yeast extract (Carl Roths GmbH, Germany), succinic acid (Unichem, China), ATP (Tris salt), CCCP, DCCD, DMSO, Ph<sub>2</sub>I (Sigma, Aldrich, USA), and the other reagents used were of analytical grade; used acid was neutralized by NaOH. The average data are represented from three independent experiments; the standard errors were calculated as described (Lakin 1992) and did not exceed 5 % if not indicated.

## Results and discussion

The effects of DMSO and Ph<sub>2</sub>I on *R. sphaeroides* growth properties

The growth characteristics were determined and pH changes were monitored during phototrophic growth of *R. sphaeroides* MDC 6521 in the presence of various concentrations of DMSO and Ph<sub>2</sub>I.

Both reagents affected the cell specific growth rate. When Ph<sub>2</sub>I in various concentrations was added into the growth medium, the cells yield was lower than that of the control, indicating that 1, 5 and 10  $\mu$ M Ph<sub>2</sub>I could inhibit the growth of these bacteria (see Fig. 1). It should

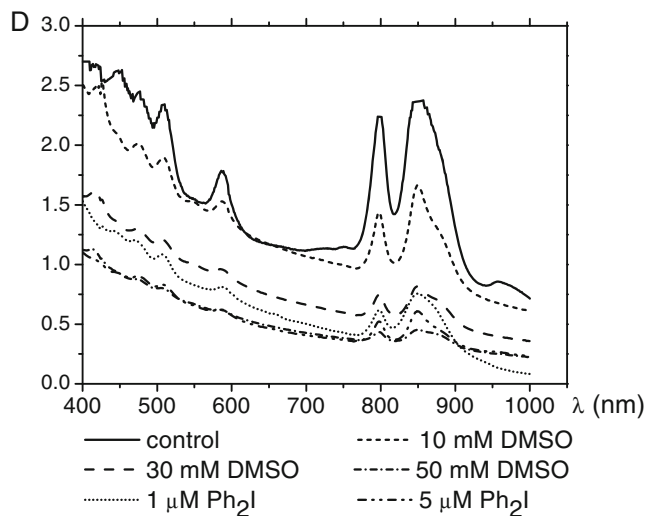


**Fig. 1** The effects of DMSO and Ph<sub>2</sub>I at various concentrations on *R. sphaeroides* MDC 6521 specific growth rates. This growth property was determined as described in Materials and methods

be noted that *R. sphaeroides* was unable to grow in medium with 20–40  $\mu\text{M}$   $\text{Ph}_2\text{I}$  (not shown). For understanding the inhibitory effect of  $\text{Ph}_2\text{I}$  was also studied was the effect of its solvent DMSO on the cell specific growth rate. In the presence of 10 mM DMSO this rate was the same as the control (no additions) (Fig. 1). The high concentration of DMSO (30–50 mM) suppressed the growth of bacterium: the specific growth rate decreased to  $\sim 3$ –4 fold in comparing with the control (see Fig. 1). These data were similar to the results on DMSO effects obtained for *E. coli* (Markarian et al. 2002).

Formation of the photosynthetic apparatus takes place during phototrophic growth of *R. sphaeroides*; it consists of two light-harvesting (LH) complexes (B800-850 and B875) surrounding a photochemical reaction center (RC) (Horne and McEwan 1998; Kim et al. 2006). The LH complexes consist of a combination of polypeptides and pigments, which include carotenoids and bacteriochlorophyll *a* (Bchl *a*) (Gabrielyan and Trchounian 2009a). As known, B875 complex accumulates and transfers the light energy to a RC (Kim et al. 2006).

For the first time the absorption spectra of *R. sphaeroides* intact cells grown with DMSO and  $\text{Ph}_2\text{I}$  was investigated in order to reveal pigments synthesis. As shown in Fig. 2, in the control cells absorption spectrum it was observed the several peaks in wavelength region of 400 to 1000 nm, which are typical for this bacterium and are indicators for their LH components (Gabrielyan and Trchounian 2012; Hakobyan et al. 2012; Horne and McEwan 1998; Paronyan 2002). These peaks could be assigned to carotenoids (450, 478 and 510 nm) and Bchl *a* (590, 800, 850 and 875 nm), respectively.



**Fig. 2** Absorption spectra of *R. sphaeroides* MDC 6521 bacterial suspension, grown in the presence of 10–50 mM DMSO and 1–5  $\mu\text{M}$   $\text{Ph}_2\text{I}$ , which is recorded as described in Materials and methods (D was optical density)

*R. sphaeroides* pigments have been found to be significantly sensitive to the content of the growth medium (see Fig. 2). By addition of DMSO and  $\text{Ph}_2\text{I}$  a decrease in the level of B800-850 and B875 complexes was observed. Absorption peaks typical for carotenoids and Bchl *a* gradually disappeared in the presence of 30–50 mM DMSO and 1–5  $\mu\text{M}$   $\text{Ph}_2\text{I}$  (see Fig. 2). Similar data were reported by Horne and McEwan (Horne and McEwan 1998), showing that the addition of DMSO caused a decrease in the level of LH complexes and a decrease in the level of carotenoids.

The effect of DMSO and  $\text{Ph}_2\text{I}$  on *R. sphaeroides* external pH,  $E_h$  and  $\text{H}_2$  production

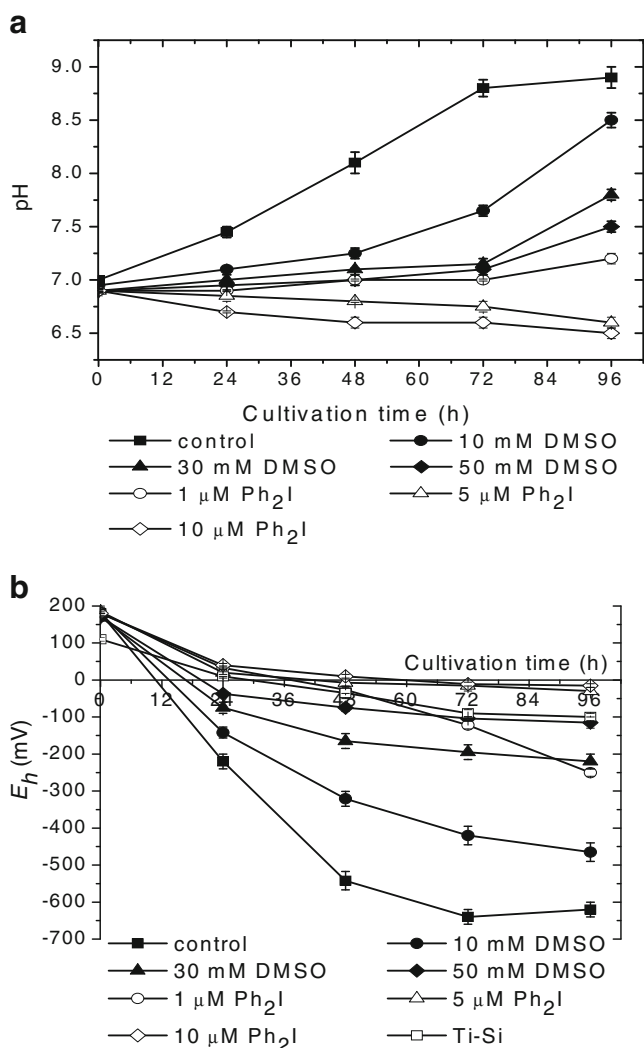
The external pH and  $E_h$  are indicative parameters for bacterial growth under various conditions (Husted et al. 1993; Vassilian and Trchounian 2009). Both reagents used affected external pH and  $E_h$  during the *R. sphaeroides* anaerobic growth upon illumination (Fig. 3). In the presence of 10 mM DMSO pH changes during bacterial growth were similar to the control (no addition): pH increased to 8.5–8.9 during the growth up to 96 h (Fig. 3a). The other kinetics of pH change were observed in the presence of  $\text{Ph}_2\text{I}$ : the pH value increased to  $\sim 7.20$  in the presence of 1  $\mu\text{M}$   $\text{Ph}_2\text{I}$ , and decreased to  $\sim 6.50$  by addition of 5–10  $\mu\text{M}$   $\text{Ph}_2\text{I}$  (Fig. 3a).

Bacterial anaerobic growth has been shown to be connected with  $E_h$  falling from positive down to low negative values (Gabrielyan et al. 2010; Hakobyan et al. 2012), indicating the enhancement of reduction processes related to the formation of fermentation end-products, the production of amino acids and the synthesis of proteins and other compounds; this is typical for the metabolic processes under anaerobic conditions (Vassilian and Trchounian 2009).

$E_h$  measured by a Pt electrode gradually decreased during the growth with 10–50 mM DMSO (Fig. 3b): potential drop was more intensive in the presence of 10 mM DMSO (up to  $-465$  mV) (Table 1). Such decrease in  $E_h$  might indicate enhanced  $\text{H}_2$  production, which also was confirmed by chemical assay (see Materials and methods). In the presence of  $\text{Ph}_2\text{I}$   $E_h$  was not changed much (see Fig. 3b). By addition of 10  $\mu\text{M}$   $\text{Ph}_2\text{I}$ ,  $E_h$  decreased less – to  $\sim 15$  mV (see Table 1).

The  $\text{H}_2$  yield of  $0.85 \text{ mmol L}^{-1}$  was obtained for *R. sphaeroides* in the medium with 10 mM DMSO, which was  $\sim 6$  fold lower than that of the control ( $5.24 \text{ mmol L}^{-1}$ ) (Table 1). The bacterium was unable to produce  $\text{H}_2$  in the presence of 30–50 mM DMSO and  $\text{Ph}_2\text{I}$ . These results indicated that DMSO could inhibit  $\text{H}_2$  production depending on its concentration.

The production of  $\text{H}_2$  and  $\text{CO}_2$  from various organic substrates by purple bacteria is observed in anaerobic conditions under illumination (Gabrielyan and Trchounian 2009a; Kapdan and Kargi 2006; Golomysova et al. 2010).



**Fig. 3** The pH (a) and  $E_h$  (b) changes of *R. sphaeroides* MDC 6521 during the anaerobic growth at DMSO and Ph<sub>2</sub>I at various concentrations. pH was measured at regular intervals by using a pH-potentiometer;  $E_h$  was measured at regular intervals using Pt and Ti-Si electrodes (see Materials and methods)

**Table 1** The effects of DMSO and Ph<sub>2</sub>I at various concentrations on H<sub>2</sub> yield of *R. sphaeroides* MDC 6521 during growth up to 96 h in phototrophic anaerobic conditions and its correlation with  $E_h$

Reagents	H <sub>2</sub> yield <sup>a</sup> (mmol L <sup>-1</sup> )	Final $E_h$ (mV)
Control (no addition)	5.24	-620±20
DMSO (10 mM)	0.85	-465±25
DMSO (30 mM)	–	-220±20
DMSO (50 mM)	–	-115±15
Ph <sub>2</sub> I (1 μM)	–	-250±10
Ph <sub>2</sub> I (5 μM)	–	-30±10
Ph <sub>2</sub> I (10 μM)	–	-15±5

<sup>a</sup> The mean values calculated by decrease in  $E_h$  (see Materials and methods) are represented

Minus (-) sign represented absence of H<sub>2</sub> production

The substrate is oxidized in the tricarboxylic acid cycle with the derived electron passing through an electron transfer pathway via NAD/NADH and ferredoxin, before being recombined with proton by a nitrogenase to produce H<sub>2</sub> (Golomysova et al. 2010). This process is thermodynamically uphill and requires ATP hydrolysis, which is produced by the photosynthetic proton gradient. High concentration of DMSO might increase an interaction with CO<sub>2</sub>, decreasing nitrogenase activity, whereas Ph<sub>2</sub>I affects the formation of photosynthetic pigments and hydrogenase synthesis.

The effects of Ph<sub>2</sub>I on intracellular pH and  $\Delta\psi$  in *R. sphaeroides*

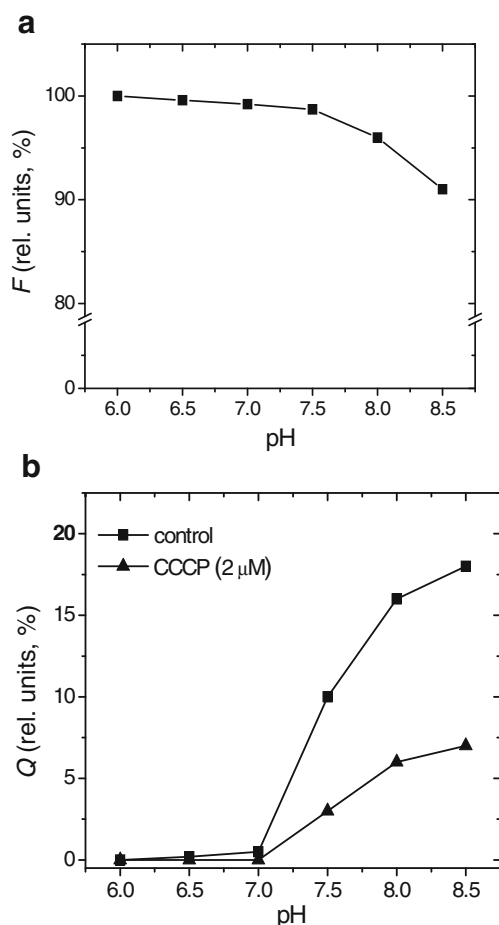
The effect of DMSO and Ph<sub>2</sub>I on growth properties and H<sub>2</sub> production observed under anaerobic conditions may be explained by the variation of the bioenergetics characteristics and ATP yields for these conditions.

Two components of  $\Delta p - \Delta pH$  and  $\Delta\psi$  were measured in *R. sphaeroides*, grown in anaerobic conditions under light. The distribution of 9-AA between external and intracellular spaces in the bacterial cells reflects the pH gradient across the cytoplasmic membrane. Dependence of 9-AA fluorescence on the medium pH without bacteria is presented in Fig. 4a: the intensity of fluorescence remained constant and decreased insignificantly at pH below 7.50. The fluorescence quenching by addition of *R. sphaeroides* occurred when [pH]<sub>out</sub> was higher than [pH]<sub>in</sub> (Fig. 4b). The [pH]<sub>in</sub> measured by the 9-AA quenching was 7.00±0.05 (Hakobyan et al. 2011). The fluorescence quenching was eliminated by addition of 2 μM CCCP (Fig. 4b). The internal pH was not sensitive to Ph<sub>2</sub>I addition (not shown).

At pH 7.50 the  $\Delta\psi$  in *R. sphaeroides* was -98 mV (Table 2).  $\Delta\psi$  was compensated for by a reversed  $\Delta pH$  of +30 mV (Hakobyan et al. 2011), resulting in the  $\Delta p$  of -68 mV (Table 2). The addition of succinate (30 mM) resulted in an increase of  $\Delta\psi$  to -120 mV. In the presence of protonophore CCCP  $\Delta\psi$  was ~2 fold lower compared with the control (Table 2). The F<sub>0</sub>F<sub>1</sub>-ATPase inhibitor DCCD did not affect  $\Delta\psi$ . The addition of Ph<sub>2</sub>I decreased  $\Delta\psi$  at concentrations of 20 μM and higher; lower concentrations of Ph<sub>2</sub>I have no valuable effect on  $\Delta\psi$ , as well as Ph<sub>2</sub>I solvent DMSO (Table 2).

The effect of Ph<sub>2</sub>I on ATPase activity of *R. sphaeroides* membrane vesicles

*R. sphaeroides* MDC 6521 membrane vesicles demonstrated ATPase activity, which was inhibited by various inhibitors (Hakobyan et al. 2011; Gabrielyan and Trchounian 2009b). The incubation of membrane vesicles of this bacterium with Ph<sub>2</sub>I caused at marked inhibition in ATPase activity: the inhibition revealed concentration dependence (Fig. 5). The inhibition in the ATPase activity by the concentrations of



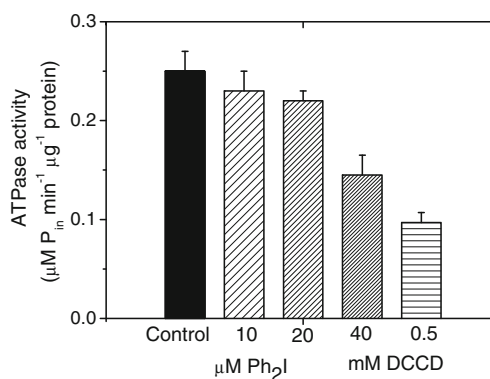
**Fig. 4** pH dependence of 9-AA fluorescence (a) and its quenching (b) induced *R. sphaeroides* MDC 6521. The fluorescence intensity ( $F$ ) in relative units at pH 6.0 was taken as 100 %, the fluorescence quenching ( $Q$ ) was expressed in % of the initial value of  $F$ . 2  $\mu\text{M}$  CCCP added eliminated the quenching

$\text{Ph}_2\text{I}$  lower than 20  $\mu\text{M}$  did not overreach 10 %. Valuable inhibition in the ATPase activity was observed in the presence of 40  $\mu\text{M}$   $\text{Ph}_2\text{I}$  (~40 %). DMSO did not affect the ATPase activity, whereas the  $\text{F}_0\text{F}_1$ -ATPase inhibitor DCCD

**Table 2** Effect of various reagents on the membrane potential and proton motive force in *R. sphaeroides* MDC 6521

Reagents	$\Delta\psi$ (mV)	$\Delta p^a$ (mV)
Control (no additions)	$-98 \pm 7$	$-68 \pm 7$
Succinate (30 mM)	$-120 \pm 3$	$-90 \pm 3$
CCCP (2 $\mu\text{M}$ )	$-57 \pm 5$	$-27 \pm 5$
DCCD (0.5 mM)	$-105 \pm 7$	$-30 \pm 7$
DMSO (10–50 mM)	$-96 \pm 6$	$-66 \pm 6$
$\text{Ph}_2\text{I}$ (1 $\mu\text{M}$ )	$-96 \pm 7$	$-66 \pm 7$
$\text{Ph}_2\text{I}$ (10 $\mu\text{M}$ )	$-90 \pm 2$	$-60 \pm 2$
$\text{Ph}_2\text{I}$ (20 $\mu\text{M}$ )	$-70 \pm 4$	$-40 \pm 4$

<sup>a</sup>  $\Delta p$  calculated as a sum  $\Delta\psi$  and reversed  $\Delta\text{pH}$  of +30 mV



**Fig. 5** The ATPase activity of *R. sphaeroides* strain MDC 6521 membrane vesicles in the presence of 10–40  $\mu\text{M}$   $\text{Ph}_2\text{I}$ . The ATPase activity was calculated by colorimetric determination of inorganic phosphate ( $\text{P}_{\text{in}}$ ) liberation per time and protein upon adding ATP adding (see Materials and methods)

suppressed the ATPase activity by 60 %. The results obtained confirm the role of photosynthetic pigments and hydrogenase but not ATPase in the  $\text{H}_2$  production process in this bacterium.

### Concluding remarks

*R. sphaeroides* have been shown to grow well in anaerobic conditions at 30  $^{\circ}\text{C}$  under illumination, using succinate as carbon source and yeast extract as nitrogen source (Hakobyan et al. 2012). During bacterial photosynthesis the carbon source—organic acid can be completely decayed up to 2H and  $\text{CO}_2$  in anaerobic conditions whereas nitrogen source is used for synthesis of nucleic acids and proteins (Gabrielyan and Trchounian 2009a; Kapdan and Kargi 2006; Golomysova et al. 2010). During electron transport by photosynthesis,  $\text{H}^+$  is pumped through the membrane,  $\text{H}^+$ -gradient is generated, and the proton-translocating  $\text{F}_0\text{F}_1$ -ATPase uses it to make ATP (Golomysova et al. 2010). The latter serves as energy source for many processes but it is important that ATP is used for the transfer of electrons to ferredoxin. At the same time, electrons and  $\text{H}^+$  form  $\text{H}_2$  via hydrogenase or nitrogenase but appropriate mechanisms and regulation are not clear. Thus,  $\text{H}_2$  production by *R. sphaeroides* is interesting in two aspects at least: (1) which enzyme—nitrogenase or hydrogenase is responsible for  $\text{H}_2$  production depending on the conditions, (2) what is a role of the  $\text{F}_0\text{F}_1$ -ATPase.

The hydrogenases known for purple non sulfur bacteria belong to the class of [Ni-Fe]-hydrogenases which are suggested to act as  $\text{H}_2$ -uptaking or oxidizing hydrogenases (Basak and Das 2007; Das et al. 2006; Colbeau et al. 1993). However,  $\text{H}_2$ -uptaking hydrogenases frequently coexist with  $\text{H}_2$ -producing hydrogenases or nitrogenases and can recycle  $\text{H}_2$  within bacteria, greatly decreasing overall  $\text{H}_2$  yield (Gabrielyan and Trchounian 2009a). Reversibility of

hydrogenases is suggested for different bacteria (Trchounian et al. 2011b; Trchounian et al. 2012) so the hydrogenase in *R. sphaeroides* might be responsible for H<sub>2</sub> production. The results with inhibitory effects of DMCO and Ph<sub>2</sub>I on H<sub>2</sub> production and ATPase activity in *R. sphaeroides* provide novel evidence for involvement or responsibility of hydrogenase and DCCD-inhibited F<sub>0</sub>F<sub>1</sub>-ATPase in H<sub>2</sub> production by this bacterium. Moreover, these data indicate their involvement or role in  $\Delta p$  generation and utilization. This is likely with an idea about a relationship between hydrogenase and the F<sub>0</sub>F<sub>1</sub>-ATPase in H<sub>2</sub> production and H<sup>+</sup> transport by *R. sphaeroides* (Gabrielyan and Trchounian 2009b). Moreover, this could mean that *R. sphaeroides* hydrogenase has a role in regulation of pH like [Ni-Fe]-hydrogenases in other bacteria (Trchounian et al. 2011b; Trchounian et al. 2012). Besides, DMSO might increase an interaction of nitrogenase with CO<sub>2</sub>, decreasing nitrogenase activity and affecting H<sub>2</sub> production.

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